# A new foundation for Statistical Physics Entaglement and the Second Law

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Thermodynamics

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Statistical Mechanics

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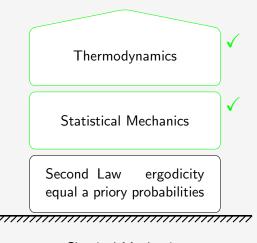
Statistical Mechanics

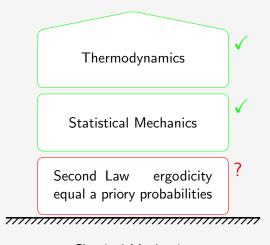
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Thermodynamics

Statistical Mechanics

Second Law ergodicity equal a priory probabilities





Clausius

Heat generally can not spontaneously flow from a material at lower temperature to a material at higher temperature.

[1,en.wikipedia.org]

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Kelvin, Planck, Kinzel

It is impossible to convert heat completely into work in a cyclic process.

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Boltzmann (H-Theorem)

The entropy in a closed system can not decrease. It stays constant only for reversible processes.

[1,en.wikipedia.org]

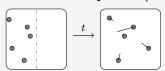
#### H-Theorem:

Thus, one may prove that, because of the atomic movement in systems consisting of arbitrarily many material points, there always exists a quantity which, due to these atomic movements, cannot increase, and this quantity agrees, up to a constant factor, exactly with the value [of] the well-known integral  $\int \frac{\delta Q}{T}$ .

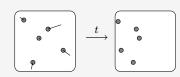
This provides an analytical proof of the Second Law [...] we immediately reach the result that  $\int \frac{\delta Q}{T}$  is in general negative and zero only in a limit case.

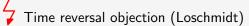


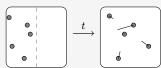
## Time reversal objection (Loschmidt)



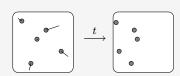








$$\vec{v} \rightarrow -\vec{v}$$



Recurrence objection (Ponicaré)

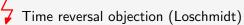


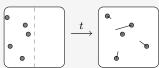
$$\xrightarrow{t} \dots \xrightarrow{t}$$



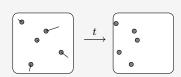
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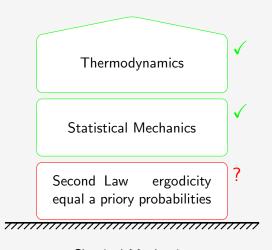






Recurrence objection (Ponicaré)

⇒ We need a probabilistic H-Theorem! (Ehrenfest)

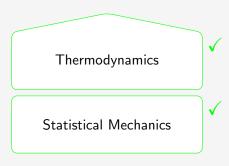


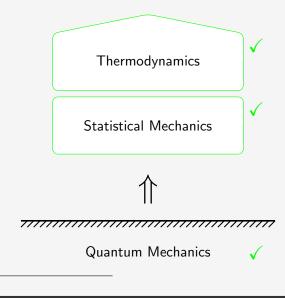
Classical Mechanics

[2, 3]

Thermodynamics Statistical Mechanics Second Law ergodicity equal a priory probabilities Classical Mechanics

[2, 3]





#### Pure state quantum Statistical Mechanics . . .

- ... must be capable of:
  - reproducing results obtained from ensemble averages
  - explaining equilibration
  - explaining initial state independence
  - . . . .

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Technical introduction

#### ■ Pure Quantum Mechanics

$$|\psi\rangle \in \mathcal{P}(\mathcal{H})$$

$$\langle \psi | \psi \rangle = 1$$

$$|\psi_t\rangle = U_t |\psi_0\rangle$$

$$A = A^{\dagger}$$

$$\langle A \rangle_{\psi} = \langle \psi | A | \psi \rangle$$

$$U_t = e^{-i \mathcal{H} t}$$

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mixtures: 
$$\rho = p \psi_1 + (1-p) \psi_2$$

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■ Time average

$$\omega = \langle \rho_t \rangle_t = \lim_{\tau \to \infty} \frac{1}{\tau} \int_0^\tau \rho_t \, dt$$

### What is a random state?

#### Parametrization

$$|\psi\rangle = \sum_{i} c_{i} |i\rangle$$
  $\langle i|j\rangle = \delta_{ij}$  
$$1 = \sum_{i} |c_{i}|^{2}$$
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### What is a random state?

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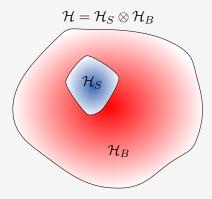
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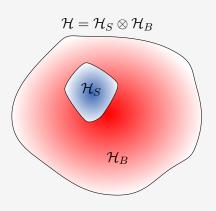
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#### Haar measure

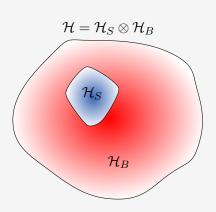
- **2** Random U;  $|\psi\rangle = U |0\rangle$





$$\rho \in \mathcal{M}(\mathcal{H})$$

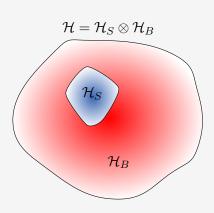
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 $reduced\ state \rightarrow locally\ observable$ 

# A very weak assumption

In the following we assume non-degenerate energy gaps.

#### Definition

A Hamiltonian has non-degenerate energy gaps iff:

$$E_k - E_l = E_m - E_n$$

$$\implies k = l \land m = n \text{ or } k = m \land l = n$$

Equilibration

#### Lemma 1

For every  $\psi_0 \in \mathcal{P}(\mathcal{H})$ 

$$\langle \mathcal{D}(\rho_t^S, \omega^S) \rangle_t \le \frac{1}{2} \sqrt{\frac{d_S^2}{d^{\text{eff}}(\omega)}}$$

where

$$\rho_t^S = \operatorname{Tr}_B \psi_t \qquad \qquad \omega^S = \langle \rho_t^S \rangle_t \qquad \qquad \omega = \langle \psi_t \rangle_t$$

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 $\Longrightarrow \rho_t^S$  equilibrates if  $d^{\mathrm{eff}}(\omega)$  is large.

#### Lemma 2

For random  $\psi_0 \in \mathcal{P}(\mathcal{H})$ 

$$\Pr\left\{d^{\text{eff}}(\omega) < \frac{d}{4}\right\} \le 2 e^{-c\sqrt{d}}$$

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# Equilibration is generic

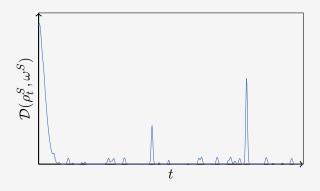
Lemma 1 + Lemma 2

If d is large  $\rho_t^S$  equilibrates for almost all initial states  $\psi_0$ .

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Lemma 1 + Lemma 2

If d is large  $\rho_t^S$  equilibrates for almost all initial states  $\psi_0$ .



Initial state independence

# Entanglement produces disorder

#### Lemma 3

If almost all energy eigenstates are close to locally completely mixed

$$\operatorname{Tr}_B |E_k\rangle\langle E_k| \approx \frac{1}{d_S} \mathbb{1}_{d_S \times d_S}$$

almost all equilibrium states are close to locally completely mixed

$$\omega^S = \langle \rho_t^S \rangle_t \approx \frac{1}{d_S} \mathbb{1}_{d_S \times d_S}.$$

# Random Hamiltonians are locally mixed

#### Lemma 4

Almost all eigenstates  $|E_k\rangle$  of a random Hamiltonian on  $\mathcal{H}$  with  $d_B\gg d_S$  are close to locally completely mixed in the sense that:

$$\Pr\left\{\forall k : \mathcal{D}(\operatorname{Tr}_{B}|E_{k})\langle E_{k}|, \frac{1}{d_{S}}\mathbb{1}_{d_{S}\times d_{S}}) \leq \frac{\epsilon}{d_{S}}\right\}$$
$$\geq 1 - 2d_{S}d_{B}\left(\frac{10d_{S}}{\epsilon}\right)^{2d_{S}} e^{-Cd_{B}\epsilon^{2}}$$

Toward a probabilistic H-Theorem

Lemma 1 subsystems equilibrate  $\Leftarrow$  effective dimension is high

Lemma 1 subsystems equilibrate ← effective dimension is high Lemma 2 random states have a high effective dimension

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Lemma 3 initial state independence ← locally mixed energy eigenstates
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Lemma 1 subsystems equilibrate ← effective dimension is high

Lemma 2 random states have a high effective dimension

Lemma 3 initial state independence ← locally mixed energy eigenstates

Lemma 4 random Hamiltonians have locally mixed energy eigenstates
```

### What about the Second Law?

### Probabilistic pseudo quantum H-Theorem

Almost all initial states of a large Quantum system of dimension d are such that under the time development induced by a generic Hamiltonian the states of all small subsystems with dimension  $d_S \ll d$  are close to an equilibrium state for almost all times.

The equilibrium state is independent of the initial state and maximizes the local Von Neumann entropy.

■ Boltzmann's H-Theorem is about closed systems!

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All I want to say is that:

There is a generic tendency to maximize entropy.

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Thank you for your attention!